

## SYNTHESIS AND EVALUATION OF SOME NEW 6-FLURO-QUINOLIN-4 (1H)-ONE DERIVATVES FOR THEIR ANTI-MICROBIAL ACTIVITIES

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### ABSTRACT

The present work is a bonafide and novel for the synthesis of 6-fluro-quinolin-4(1H)-one derivatives. Around 30 new derivatives were synthesized, with the standard chemicals and well established procedures. The synthesized compounds were tested for their preliminary tests, physical constants, TLC, solubility, etc. IR, <sup>1</sup>H-NMR Spectra and CHN analysis confirmed the structures of the final compounds. The proposed compounds were screened for their antimicrobial, antifungal, antitubercular and anti-inflammatory activities with the standard drugs in the well-equipped microbiology and pharmacology lab by using standard methods.

**Key Words.** Antibacterial, Antifungal, 6-fluro-quinolin-4 (1H)-one derivatives

### INTRODUCTION

The need of new anti-microbial agents is justified because more microorganisms are being resistance to the present drugs available in the market. World wide researchers are trying to synthesize new drugs with better pharmacokinetic and dynamic properties with less adverse effects. The literature survey suggests that the 6-fluro-quinolin-4(1H)-one have proved to be good bioactive molecules. They have shown diverse biological activities like anti-bacterial, anti-fungal, anti-inflammatory, anti-tubercular, anticonvulsant, anti-HIV, cardiac stimulant, diuretic and anticancer etc. Therefore in view of above facts it was thought of interest to synthesize some 6-fluro-quinolin-4 (1H)-one Derivatives. IR, <sup>1</sup>H-NMR Spectra and CHN analysis confirmed the structures of the final compounds. The proposed compounds were screened for their antimicrobial, antifungal, antitubercular and anti-inflammatory activities with the standard drugs in the well-equipped microbiology and pharmacology lab by using standard methods

### MATERIALS AND METHODS:

#### EXPERIMENTAL:

Melting points were determined in open capillary method and are uncorrected. Purity of the compound was checked on Silica gel TLC plates. IR spectra were recorded on Thermo Nicolet IR 200 spectrophotometer using KBr disc method. <sup>1</sup>H-NMR spectra were recorded on Bruker AMX-400, DMSO d<sub>6</sub> as internal standard. Combustion analyses were found to be within the limits of permissible errors.

#### A] Preparation of diethyl 2-((3-chloro-4-fluorophenylamino) methylene) malonate (I<sub>1</sub>). [5]

A mixture of 3-chloro-4-fluro-aniline (0.01mol) and Diethyl ethoxy methylene malonate (0.01mol) was heated at 120-130<sup>0</sup>C for two hours the resulting ethanol was evaporated off. The crude solid was filtered, dried and recrystallized from n-hexane. M.P. 54-55<sup>0</sup>C.

#### B] Preparation of ethyl 7-chloro-6-fluoro-4-oxo-1, 4-dihydroquinoline-3-carboxylate (I<sub>2</sub>) [6].

Diphenyl ether was heated under stirring at 240<sup>0</sup>C. 0.158 mol of ethyl anilinomethylene malonate was added slowly to the boiling diphenyl ether for about 15 minutes after adding the mixture

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was refluxed in oil bath for two hours. The mixture was cooled, filtered and washed twice with 200 ml pet ether. The crude solid obtained was dried and purified by recrystallization twice from DMF.M.P. 270<sup>0</sup>C.

**C] Preparation of 7-chloro-6-fluro-4-oxo-1,4-dihydroquinoline-3-carbohydrazide (I<sub>3</sub>). [7]**

Ethyl 7-chloro-6-fluro-4-oxo-1,4-dihydroquinoline-3-carboxylate I<sub>2</sub> (0.035 mol) in ethanol (20 ml), DMF (10 ml) was added to 99% hydrazine hydrate (0.035mol) and was refluxed for 12 hours. Excess solvent was removed by distillation and the mixture was poured into crushed ice. The solid separated was filtered, washed with water and dried. The crude solid was purified by recrystallized from ethanol dioxan mixture (1.1) to give whitish-brown solid ;M.P.245-247<sup>0</sup>C.

**D] Preparation of 7-chloro-6-fluro-3-(3-methyl-5-oxo-4, 5 dihydropyrazole-1-carbonyl) quinolin-4(IH)-one (I<sub>4</sub>). [7]**

7-chloro-6-fluro-4-oxo-1, 4-dihydroquinoline-3-carbohydrazide I<sub>3</sub> (0.018 mol) in ethanol (20 ml) was added to ethyl acetoacetate (0.02 mol) and refluxed for 4 hours. To this mixture 2 ml of acetic acid was added and further refluxed for two hours. Excess of solvent was removed and the mixture was poured into ice water. The solid separated was filtered, washed with water and dried. It was purified by recrystallized from ethanol to afford whitish amorphous solid; M. P 218-220<sup>0</sup>C.

**E] Preparation of 6-fluro-3- (3-methyl-5-oxo-4, 5-dihydro- H-pyrazole-1-carbonyl)-7-(substituted) quinolin-4 (1H)-one (V<sub>1</sub>-V<sub>5</sub>). [8]**

The mixture of 7-chloro-6-fluro-3- (3-methyl-5-oxo-4, 5 dihydropyrazole-1-carbonyl) quinolin-4(IH)-one (0.005 mole), Piperazine (0.01 mole), Pyridine (10 ml) and Triethyl amine (3 ml) was stirred at 120-130<sup>0</sup>C for 10 hours. After

completion of the reaction, the reaction mixture was cooled to room temperature. The mixture was poured into crushed ice and neutralized with dilute HCl. The solid product was filtered dried and recrystallized from DMF and Ethanol (2.1). Similarly V<sub>2</sub>, V<sub>3</sub>, V<sub>4</sub> and V<sub>5</sub> were prepared by using Morpholin, Imidazole, Piperadine and Pyrollidine. Analytical data's were given in the table.

**F] Preparation of 7-chloro-6-fluro-3-(5-substituted-1, 3, 4-oxadiazole-2-yl) quinolin-4(1H)-one (P<sub>1</sub>-P<sub>7</sub>). [8,9]**

A mixture of an equimolar quantity of 7-chloro-6-fluro-4-oxo-1, 4-dihydroquinoline-3-carbohydrazide I<sub>3</sub> (0.006mol) and substituted aromatic acids (0.006) in 15 ml of phosphorus oxychloride was refluxed for 8 hours. The progress of the reaction was monitored by TLC using ethyl acetate. acetone (9.1) as eluent. The reaction mixture was cooled and poured carefully on to crushed ice (200g) with constant stirring and neutralized with sodium bicarbonate solution (10%w/v). The resulting solid thus obtained was collected by filtration, washed well with cold water, dried and recrystallized from ethanol. DMF (2.1) to give (P<sub>1</sub>-P<sub>7</sub>). Analytical data was given in the table1.

**ANTIBACTERIAL ACTIVITY. [1, 2]**

The compounds were tested in-vitro for their antibacterial activity against two microorganisms viz. *Escherichia coli* (NCTC 10418), and *Staphylococcus aureus* (NCTC 6571) which are pathogenic in human beings.

**Method.** Disc Agar diffusion method using Mueller-Hinton agar using E.coli, *S. aureus*

**ANTIFUNGAL ACTIVITY. [3,4]**

The compounds were tested in-vitro for their antifungal activity against *Candida albicans* (ATCC 10231) and *Aspergillus niger* (ATCC 16).

**Method.** Cup-Plate agar diffusion method using Sabouraud-dextrose agar using *C.albicans* & *A. Niger*.

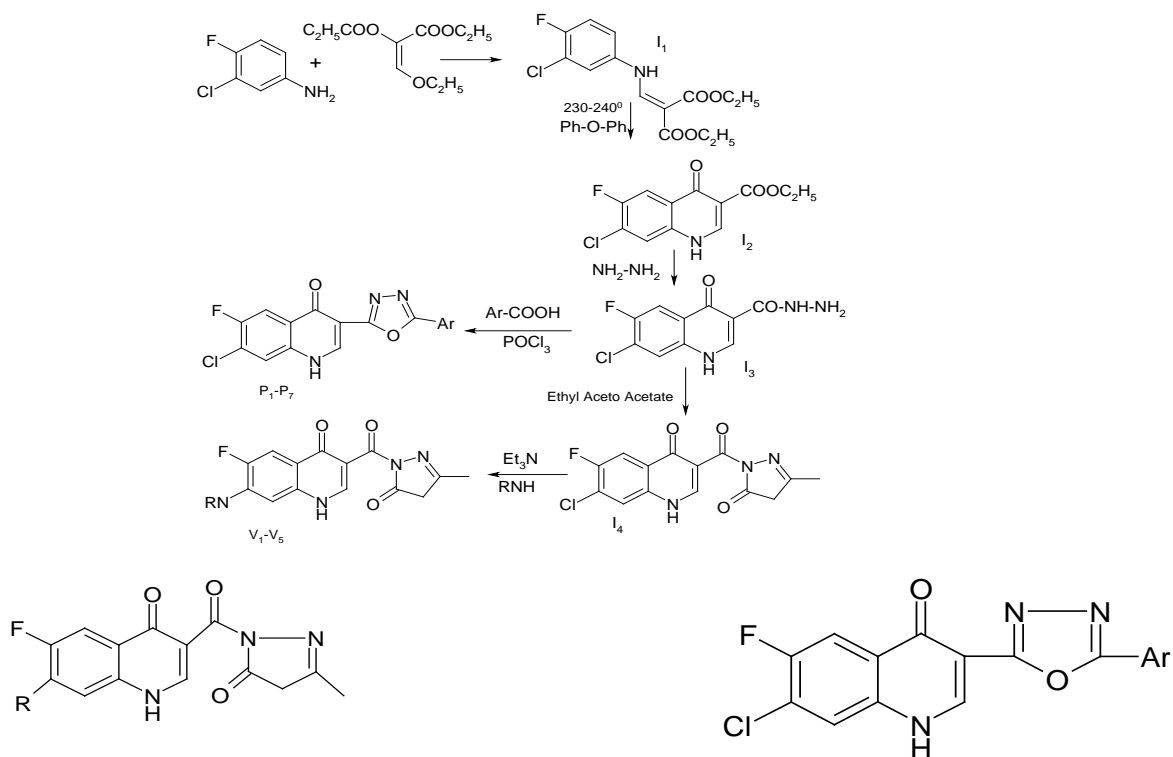
### RESULT AND DISCUSSION:

In the present research work, we have synthesized 13 new 6-fluoro-Quinolinones as explained in the scheme. The purity of the compounds was checked by TLC and melting point. Structures of these compounds were confirmed by IR, <sup>1</sup>HNMR and elemental analysis. Possible QSAR studies are carried out. The synthesized compounds were subjected to anti bacterial and anti fungal activities by Disc diffusion method against the standard

strains of *E.coli* (NCTC-10418), *S.aureus* (NCTC-6571) and for anti fungal activities, *C.albicans* (ATCC-1023), *A.niger* (ATCC-16) by cup plate method.

Compound I<sub>4</sub>, V<sub>1</sub>, V<sub>3</sub>, P<sub>1</sub>, P<sub>2</sub>, P<sub>4</sub>, P<sub>5</sub> have shown promising anti bacterial activity against Ciprofloxacin as the standard drug at 100 mcg/ml. Compound I<sub>4</sub>, V<sub>2</sub>, V<sub>5</sub>, P<sub>1</sub>, P<sub>4</sub>, P<sub>5</sub>, P<sub>7</sub> have shown excellent anti fungal activity against Griseofulvin as the standard drug at 100 mcg/ml. With the suitable molecular modification and manipulation with possible SAR studies of these compounds, promising anti microbial agents can be obtained.

### SCHEME (P<sub>1</sub>-P<sub>7</sub>, V<sub>1</sub>-V<sub>5</sub>)



Compound	-R	Compound	-Ar
V <sub>1</sub>		P <sub>1</sub>	
V <sub>2</sub>		P <sub>2</sub>	
V <sub>3</sub>		P <sub>3</sub>	
V <sub>4</sub>		P <sub>4</sub>	
V <sub>5</sub>		P <sub>5</sub>	
		P <sub>6</sub>	
		P <sub>7</sub>	

Table No. 1. Anti-bacterial and Anti-fungal activity of 6-fluro-quinolin-4(1H)-ones compounds.

Compd.	Zone of inhibition at 100µg/mL (in mm.)			
	<i>E. coli</i>	<i>S.aureus</i>	<i>A. niger</i>	<i>C. albicans</i>
I <sub>4</sub>	23	21	22	21
V <sub>1</sub>	22	23	16	18
V <sub>2</sub>	19	22	23	22
V <sub>3</sub>	24	12	18	17
V <sub>4</sub>	16	17	15	16
V <sub>5</sub>	17	18	22	23
P <sub>1</sub>	24	21	24	25
P <sub>2</sub>	23	24	17	18
P <sub>3</sub>	24	25	17	19
P <sub>4</sub>	22	24	22	23
P <sub>5</sub>	22	23	24	25
P <sub>6</sub>	15	16	18	16
P <sub>7</sub>	18	17	25	26
Ciprofloxacin	24	24	-	-
Griseofulvin	-	-	25	26

Table No. 2 Analytical data of 6-fluro-quinolin-4(1H)-one compounds

Comp.	Mol. Formula	Mol. Wt.	M.P °C	Yield %	Elemental analyses			LogP	CLogP	CMR
					Calcd.	(Found)				
					C	H	N			
I <sub>4</sub>	C <sub>14</sub> H <sub>9</sub> ClFN <sub>3</sub> O <sub>3</sub>	322.	218	67	52.27	2.82	13.06	0.65	-0.513	7.810
V <sub>1</sub>	C <sub>18</sub> H <sub>18</sub> FN <sub>5</sub> O <sub>3</sub>	371	182	78	58.22	4.89	18.86	-0.25	-1.717	9.734
V <sub>2</sub>	C <sub>18</sub> H <sub>17</sub> FN <sub>4</sub> O <sub>4</sub>	372	270	67	58.06 (58.03)	4.60 (4.57)	15.05 (15.04)	-0.03	-1.703	9.518
V <sub>3</sub>	C <sub>17</sub> H <sub>12</sub> FN <sub>5</sub> O <sub>3</sub>	353	264	66	57.79	3.42	19.82	-0.71	-1.349	9.048
V <sub>4</sub>	C <sub>19</sub> H <sub>19</sub> FN <sub>4</sub> O <sub>3</sub>	370	166	75	61.61 (61.54)	5.17 (5.14)	15.13 (15.8)	1.11	-0.321	9.829
V <sub>5</sub>	C <sub>18</sub> H <sub>17</sub> FN <sub>4</sub> O <sub>3</sub>	356	238	61	60.67 (60.28)	4.81 (4.76)	15.72 (15.58)	0.69	-0.880	9.365
P <sub>1</sub>	C <sub>17</sub> H <sub>9</sub> ClFN <sub>3</sub> O <sub>2</sub>	342	106	29	59.75	2.65	12.30	3.08	1.600	8.699
P <sub>2</sub>	C <sub>17</sub> H <sub>8</sub> ClFN <sub>4</sub> O <sub>4</sub>	387	183	60	52.80 (52.74)	2.09 (1.92)	14.49 (14.42)	--	1.393	9.311
P <sub>3</sub>	C <sub>19</sub> H <sub>11</sub> ClFN <sub>3</sub> O <sub>2</sub>	368	94	35	62.05	3.01	11.43	3.87	2.194	9.905
P <sub>4</sub>	C <sub>17</sub> H <sub>10</sub> ClFN <sub>4</sub> O <sub>2</sub>	357	233	63	57.24	2.83	15.71	2.28	0.643	9.068
P <sub>5</sub>	C <sub>17</sub> H <sub>10</sub> ClFN <sub>4</sub> O <sub>3</sub>	373	258	44	54.78 (54.66)	2.70 (2.64)	15.03 (14.87)	1.89	-0.037	9.221
P <sub>6</sub>	C <sub>17</sub> H <sub>9</sub> ClFN <sub>3</sub> O <sub>3</sub>	358	100	58	57.08	2.54	11.75	2.69	0.919	8.852
P <sub>7</sub>	C <sub>17</sub> H <sub>10</sub> ClFN <sub>4</sub> O <sub>2</sub>	357	138	38	57.24 (57.18)	2.83 (2.80)	15.71 (15.67)	2.28	0.643	9.068

The combustion analysis of compounds synthesized was found to be within the limits of permissible errors.

**Table No. 3. SPECTRAL DATA**

Compound	IR (KBr) (cm <sup>-1</sup> )	<sup>1</sup> H NMR (DMSO) δ in ppm
V <sub>1</sub>	3385 N-H str.3109-C-H Ar., 1697-C=O str. 1550 C-N piperazine 899 C-H def.	6.80-7.10 2H, Ar-CH. 2.86-3.15, 4H piperazine. 2.28-2.48, 2H CH <sub>2</sub> pyrazolones. 1.22-1.27, 3H CH <sub>3</sub> .
V <sub>2</sub>	3432 N-H str.3057 C-H Ar. 1666-C=O str. 1174.C-O str,morpholine, 798-C-H def., 806 C-F str.	7.00-7.73, 3H, Ar-CH. 3.24-3.31, 4H, morpholine. 2.48-2.59, 2H, CH <sub>2</sub> pyrazole. 1.23- 1.28,3H, CH <sub>3</sub>
V <sub>3</sub>	3156.-N-H str. 3121 C-H Ar. str. 2988-C-N str. Imidazole 1690 -C=O str, 893-C-H def 844 C-F str.,	7.10-7.30, 2H, Ar-CH. 2.28-2.48, 2H, CH <sub>2</sub> imidazole. 1.20-1.25, 3H, CH <sub>3</sub> .
V <sub>4</sub>	3106 N-H str.2989 C-H Ar. str. 2857 C-N str. piperidine, 1699-C=O str. 808-C-H def. 787 C-F str.	12.41 1H, N-H. 7.93-7.96 8H, piperidine 6.18-7.10 3H, Ar-CH. 2.4-2.7 2H ,CH <sub>2</sub> pyrazole. 1.23-1.28 3H ,CH <sub>3</sub> .
V <sub>5</sub>	3284- N-H str.,3049-C-H Ar. Str, 2920 C-N str. Pyrrolidine 2852 C-H Alkyl 1678 C=Ostr. 1256 C-N Ar. str.	13.24 1H, N-H. 7.37-7.92 3H Ar-CH. 3.74-3.98 8H, pyrrolidine. 1.22-1.27 3H CH <sub>3</sub>
P <sub>1</sub>	3343 N-H str. 3102 C-H Ar. Str, 2911 C=N Ar str. 1628 C=O str., 1007 C-O ether str.	10.70 1H, NH. 7.80-7.91 1H, quinolone. 7.80-7.90 7H, Ar-CH.
P <sub>2</sub>	3123 N-H str,3000 C-H Ar. str, 2918 C=N Ar str. 1696 C=O str. 1539 N=O str 1274 C-F str,	10.91H, NH. 7.87-7.91 3H, quinolone. 6.97-7.37 3H, Ar-CH
P <sub>3</sub>	3451 N-H str, 3071 C-H Ar. str, 2866 C=N Ar str.1697 C=O str, 1259 C-F str.	10.80 1H, NH 7.80-7.91 1H, quinolone 7.20-7.37 7H, Ar-CH.
P <sub>4</sub>	3422 N-H str.,3310 C-H Ar. str, 2928 C=N Ar str.1585 C=O str. 1240 C-F str.684 C-Cl str.	12.70 1H, N-H 8.10-8.30 1H, quinolone. 6.37-6.54 7H, Ar-CH.
P <sub>5</sub>	3437 N-H str, 3387 C=N Ar 3063 C-H Ar. str.-, 3003 OH str, 1586 C=O str 1086 C-O ether str.	13.78 1H, N-H. 8.14-8.17 3H, quinolone. 9.94 1H, OH 6.41-7.12 3H, Ar-CH. 5.03 2H, NH <sub>2</sub>
P <sub>6</sub>	3304 N-H str, 3113 OH str 3053 C-H Ar. str, 3011 C=N Ar, 1665 C=O str,1138 C-O ether str, 675 C-Cl str.	10.70 1H, NH 9.80 1H OH. 7.80-7.91 1H, quinolone. 7.80-7.90 7H, Ar-CH.
P <sub>7</sub>	3427 N-H str,3061 C-H Ar. str, 2925 C=N Ar str. 1703 C=O str 1227 C-O ether str, 758 C-F str. 691 C-H def.	13.86 1H NH. 7.93-8.16 3H,quinolone. 7.35-7.37 4H, Ar-CH. 4.98 2H, NH <sub>2</sub> .

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